Numerical solution of heat and mass transfer in MHD flow in the presence of chemical reaction and Arrhenius heat generation of a stretched vertical permeable membrane

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Abstract

We present a magnetohydrodynamic flow of a uniformly stretched vertical permeable surface undergoing Arrhenius heat reaction. It is shown that the temperature, concentration and the velocity fields depend on the chemical reaction parameter. The values of temperature field increase as the order of the reaction increases, while that of velocity field decreases as the order reaction increases. Moreover, the reactant field decreases faster as we move away form the wall as we increase the reaction parameter. This paper also shows that the temperature field and reacting layers get thinner as the heat deposit Q per unit mass increases while the velocity field and the boundary layer get thinner as thermal Grashof number increases. We also show that magnetic induction and cooling of the plate (thermal Grashof number Grt>0) lowers the velocity field.

Keywords: Permeable surface, source term, thermal expansion, suction, ambient, magnetic induction.

Classification: 76W05

1.0 Introduction

Magneto - hydrodynamic (MHD) flow of electrically conducting fluids in the presence of magnetic field is encountered in many problems in geophysics and astrophysics, Ayeni R. O. Okedoye A. M. Balogun F. O. and Ayodele T. O. (2004) [1], Chamkha (2003) [2] and the literature cited therein. Also many industrial processes involve fluid flow, heat and mass transfer in the boundary layers induced by a surface moving with a uniform velocity.

Nomenclatures

c: concentration	Pr : Prandtl number	T_w : wall temperature
c_p : specific heat	Sc : Shmidt number	v: fluid transverse velocity
C: fluid concentration	E: activation energy	v : sustion valoaity
C_w : wall concentration	g : gravitational pull	V_W . suction velocity
α_D : mass diffusion-coefficient	Bo : magnetic induction	x: axial or vertical coordinate
D: chemical reaction parameter	B 0. magnetic muticition	y: transverse coordinate
Q_0 : heat generation coefficient	T: fluid temperature	Greek Symbols
	<i>n</i> : reaction order	γ : chemical reaction parameter
I v u: Nuselt number	<i>K</i> : universal gas constant	$\boldsymbol{ ho}$: fluid density

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T_w : wall temperature v: fluid transverse velocity	Greek Symbols γ : chemical reaction parameter ρ : fluid density	$oldsymbol{eta}_{ au}$: coefficient of thermal expansion $oldsymbol{eta}_c$: coefficient of mass expansion
v_W : suction velocity x: axial or vertical coordinate y: transverse coordinate	θ : dimensionless temperature R_i : Richardson number σ : electric conductivity	Subscripts W : condition on the wall ∞ : ambient condition

Chamkha (2003) [2] examined the boundary layer of an MHD flow when the heat generation is linear in temperature. Ayeni et al (2004) [1] extended the problem posed in Chamkha (2003) [2] to heat generation that is quadratic in temperature. Sakiads, (1996) studied the boundary – adjacent to a continuous moving surface. He obtained solutions by approximate and exact methods of momentum boundary – layer equations, with no heat transfer on flat and cylindrical surfaces. The corresponding heat transfer problems were considered experimentally by Griffin and Throne (1967). Vajravelu and Hadjinicolaou (1990) [5], reported on convective heat transfer in an electrically conducting fluid at a stretching surface with uniform free stream. Other examples of studies dealing with hydro – magnetic flows can be found in the papers by Gray (1980). Michiyoshi et al. (1976) and Funmizawa (1979) [4].

Okedoye and Ayeni (2007) [6] examined the problem of MHD and, mass and heat transfer when the chemical reaction is of order zero. Since most chemical reaction in conducting fluid is of order > 0, the present work is more appropriate. In this paper we extend the problem posed by Chamkha to Arrhenius heat generation and chemical reaction of order n, so that previous cases of heat generation become special cases of the present paper.

2.0 Mathematical formulation

Consider coupled heat and mass transfer by hydro – magnetic flow of a continuously moving vertical permeable surface in the presence of suction, heat generation/absorption effects, transverse magnetic field effect and Arrhenius reactions. The flow is assumed steady, laminar and two – dimensional and the surface is maintained at a uniform temperature and the concentration species, and is assumed to be infinitely long. It is also assumed that the applied transverse magnetic Reynolds number is small so that the induced magnetic field is neglected. In addition, there is no applied electric field and all of the Hall effect, viscous dissipation and Joule heating are neglected, thermo - physical properties are assumed constant except the density in the buoyancy terms of the momentum equation which is approximated according to the Boussinesq approximation.

With these assumptions, the steady equations that describe the physical situation are given as

$$\frac{\partial v}{\partial y} = 0$$
 (2.1)

$$v\frac{\partial u}{\partial y} = v\frac{\partial^2 u}{\partial^2 y} + g\beta_{\tau} \left(T - T_{\infty}\right) + g\beta_{c} \left(c - c_{\infty}\right) - \frac{\sigma B_{0}^{2} u}{\sigma}$$
(2.2)

$$pc_{p} v \frac{\partial T}{\partial y} = k \frac{\partial^{2} T}{\partial y^{2}} + Q (c - c_{\infty})^{n} e^{-E/RT}$$
(2.3)

$$v\frac{\partial u}{\partial y} = D \frac{\partial^2 C}{\partial y^2} - \gamma (c - c_{\infty})^n e^{-E/RT}$$
(2.4)

where y is the horizontal or transverse coordinate, it is the axial velocity, v is the transverse velocity, T is the fluid temperature, c is the concentration, $T \infty$ is the ambient temperature $c \infty$ is

Journal of the Nigerian Association of Mathematical Physics, Volume 11 (November 2007), 111 – 118 Solution of a stretched vertical permeable membrane, A. M. Okedoye, R. O. Ayeni J. of NAMP the ambient concentration, and ρ , *g*, *B_T*, $\beta_c v \sigma$, *B_c*, *Q*, *D*, γ and *n* are the density, gravitational acceleration, coefficient of thermal expansion, coefficient of concentration expansion, kinematics viscosity, fluid electrical conductivity, magnetic induction, heat generation/coefficient and the chemical reaction parameter and real number respectively. With the physical boundary conditions

$$u(0) = u_{W}, v(0) = -v_{W}, T(0) = T_{W}, C(0) = C_{W}$$

$$as \ y \to \infty, u \to \infty, T \to \infty, C \to \infty,$$
(2.5)

where u_w , $v_w >0$, T_w and c_w are surface velocity, suction velocity, surface temperature and concentration respectively.

3.0 Method of solution

We use the non – dimensional variables

$$y^{1} = yv_{W} / v, u^{1} = \frac{u}{u_{W}}, \ \theta = (T - T_{\infty}) \frac{E}{RT_{\infty}^{2}}, \ c^{1} = \frac{c - c_{\infty}}{c_{W} - c_{\infty}}$$
 (3.1)

From equation (2.1) i.e $\frac{dv}{dy} = 0$ on integrating we have v = constant

but
$$v(0) = -v_W = \text{constant}, \Rightarrow v(y) = -v_W$$
(3.2)

Substituting equations (3.1) and (3.2) and dropping ('), we have

$$\frac{d^{2}u}{dy^{2}} + \frac{du}{dy} + G_{rT}\theta + G_{rc}c - M^{2}u = 0$$
(3.3)

$$\frac{d^2\theta}{dy^2} + p_r \frac{d\theta}{dy} + b \operatorname{Pr} \phi C^n e^{\frac{\theta}{1 + \varepsilon \theta}} = 0$$
(3.4)

$$\frac{d^2c}{dy^2} + s_c \frac{dc}{dy} - KScbC^n e^{\frac{\theta}{1+\varepsilon\theta}} = 0$$
(3.5)

The dimensionless boundary conditions becomes

$$u(0) = 1, \theta(0) = 1, c(0) = 1$$

$$y \to \infty \quad u \to 0, \ \theta \to 0, \ c \to 0$$
(3.6)

where

$$G_{rT} = \frac{B_{T} v_{g} RT_{o}^{2}}{U_{w} EV_{w}^{2}}, G_{rc} = g \frac{B_{c} v(C_{w} - C_{\infty})}{U_{w} V_{w}^{2}}, P_{r} = \frac{\mu cp}{k}, S_{c} = \frac{V_{w}}{D}, M^{2} = \frac{\sigma \beta_{o} v}{\rho v_{w}^{2}},$$
$$\frac{vQ}{\mu c p v_{w}^{2}} = \frac{vQ}{\rho c_{p} v_{w}^{2}}, \frac{v\rho c_{p}}{k} = \phi Pr, \frac{vQ}{\mu c p v_{w}^{2}} \frac{RT_{0}^{2}}{E} = \varepsilon \phi Pr, b = (C_{w} - C_{\infty})^{n} \frac{E}{RT_{o}^{2}} e^{-\frac{1}{\varepsilon}},$$

We now re – write the system of equations (3.3) – (3.5) as a system of 1^{st} – order equations. Let, $y=X_1, C=X_2, \theta=X_3, u=X_4, C^1=X_5, \theta^1=X_6, u^1=X_7$ And in vector form we have

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$$\begin{pmatrix} X_1 \\ X_2 \\ X_3 \\ X_4 \\ X_5 \\ X_6 \\ X_7 \end{pmatrix}^1 = \begin{pmatrix} 1 \\ X_5 \\ X_6 \\ X_7 \\ Sc(KScX_2 - X_5)bX_2^n e^{\frac{X_3}{1 + \varepsilon x_3}} \\ -\Pr(X_6 + b\phi X_2^n \frac{X_3}{1 + \varepsilon x_3} \\ -\Pr(X_6 - aGrcX_2 - GrTX_3 - X_7) \end{pmatrix}$$

subject to

$ \begin{pmatrix} x_1(0) \\ x_2(0) \\ x_3(0) \\ x_4(0) \\ x_5(0) \\ x_6(0) \\ x_7(0) \end{pmatrix} =$	$ \begin{pmatrix} 0 \\ 1 \\ 1 \\ a_1 \\ a_2 \\ a_3 \end{pmatrix} $, where $a_i, i=1,3$	are guess values
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The essence of the above system is to transform the second order boundary value problem to a system of 1st order initial value problem. We continue to adjust the values of a_i , i=1,..3 until $u \rightarrow 0, \theta \rightarrow 0, c \rightarrow 0$, as $y \rightarrow \infty$ is satisfied. We then use Runge-Kutta of order four to solved the problem and the result for various choice of parameters were displayed graphically in figures 1-9.



Figure 1: Concentration profile for various values of k





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4.0 Discussion of results

In the analysis presented here, the parameters ϕ_0 , K, Ri, Pr and Sc are assigned values – Grt = 1, Grc=1, Pr=0.71, Sc=0.6, $\phi = -0.2$, and M=0.5 except where stated. It should be noted that K > 0 indicate a destructive chemical reaction, while K < 0 correspond to a generative chemical reaction. Also, $\phi_0 < 0$ indicate heat generation and $\phi_0 > 0$ imply heat absorption. And K=0 and $\phi_0 = 0$ indicate no chemical reaction and no heat generation/absorption effects respectively. Also Grt > 0 indicate cooling of the plate while Grt < 0 indicate heating of the plate.

We display in figures 1-9, the temperature profile, velocity profile and concentration profile for various values of the parameters. Figure 1 show that the concentration layer gets thinner as the rate of reactant consumption increases. On the other hand, in figures 2 and 3, the velocity and temperature layer increases as the rate of reactant consumption per unit mass increases. Figures 4 and 5 show the effect of magnetic induction and thermal Grashof number the velocity field. In figure 4, we show that velocity layer decreases as magnetic induction M increases, this is due to the opposing force put up in the flow by

Journal of the Nigerian Association of Mathematical Physics, Volume 11 (November 2007), 111 – 118 Solution of a stretched vertical permeable membrane, A. M. Okedoye, R. O. Ayeni J. of NAMP Lorenz force which always oppose the fluid velocity. On the other hand, figure 5 shows that the velocity layer decreases as thermal Grashof number Grt increases as a result of cooling of the channel plate. Figure 6 show that the temperature layer gets thinner as the heat deposit increases. Chemical reaction is being consumed more as the other of reaction increases. This is shown in figure 7, as we discovered that the concentration profile reduces as n increases. This effect of n on concentration field also affect the other fields – temperature and velocity fields. Figures 8 and 9 shows the effect of reaction order on velocity and temperature fields. We could see that increase in reaction order result in increase in temperature field, as more reaction is consumed the temperature of the system increases and vice – versa, while from figure 9, consumption of the chemical species reduces the strength of flow and thus increase in reaction order lowers the velocity field. The boundary layers of the three fields gets thinner as the reaction order increases.

5.0 Concluding remarks

Numerical solutions for mass and heat transfer by steady laminar flow of an electrically conducting and heat generating/absorbing fluid on a uniformly moving vertical permeable surface with buoyancy in the presence of Arrhenius heat generation/absorption and a first order chemical reaction were reported. Based on the obtained results, the following conclusions were deducted:

- (1) Maximum velocity reduces in value as both *Grt* and K increases.
- (2) A maximum concentration occurs during a generative chemical reaction.
- (3) Increases in destructive chemical reaction produces lower species concentration and vice versa.
- (4) The fluid body temperature increases with increase in heat generation ($\phi_0 < 0$), and a decreases a with a decrease in ($\phi_0 > 0$).
- (5) Velocity and temperature fields increase with increase in generative chemical reaction.
- (6) Velocity field decreases with increase in both Hartmann number M and cooling of the plate (Grt > 0).

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